

Study of the kinetics ...

S/020/61/137/006/015/020
B101/B201

formation of C_3H_4O and CO_2 takes place predominantly at lower temperatures, and a parallel-consecutive formation of CO_2 at higher temperatures. Ye. N. Popova, D. Ya. Nechiporuk, and M. V. Rybakova are thanked for their assistance. There are 1 figure, 3 tables, and 8 Soviet-bloc references.

ASSOCIATION: Institut fizicheskoy khimii im. L. V. Pisarzhevskogo Akademii nauk USSR (Institute of Physical Chemistry im. L. V. Pisarzhevskiy, Academy of Sciences, UkrSSR)

PRESENTED: December 10, 1960, by A. A. Balandin, Academician

SUBMITTED: December 9, 1960

Card 4/8

BELOUSOV, V.M.; GOROKHOVATSKIY, Ya.B.; RUBANIK, M.Ya.

Kinetics of oxidation of propylene to acrolein on a copper catalyst. Kin.i kat. 3 no.2:221-229 Mr-Ap '62. (MIRA 15:11)

1. Institut fizicheskoy khimii imeni L.V.Pisarzhhevskogo AN UkrSSR.

(Propene) (Acrolein) (Catalysts, Copper)

BELOUSOV, V.M.; RUBANIK, M.Ya.

Method of competing reactions used for studying the mechanism
of catalytic oxidation of lower hydrocarbons. Kin. i kat. 4
no.6:892-897 N-D '63. (MIRA 17:1)

1. Institut fizicheskoy khimii imeni Pisarzhevskogo AN UkrSSR.

BELOUSOV, V.M.

Competitive reactions as a method of investigating the mechanism
of heterogeneous catalysis. Ukr. khim. zhur. 30 no.1:32-39 '64.
(MIRA 17:6)

1. Institut fizicheskoy khimii imeni Pisarzhevskogo AN UkrSSR.

BELOUS, V. M.; CHIBISOV, K. V.

Luminescence studies of the role played by admixed silver centers
in the photolysis of silver halides. Dokl. AN SSSR 156 no. 1:
121-124 My '64. (MIRA 17:5)

1. Odesskoye vyssheye inzhenernoye morskoye uchilishche i
Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova.
2. Chlen-korrespondent AN SSSR (for Chibisov).

BELOUSOV, V.M.; GERSHINGORINA, A.V.

Adsorptive properties and reactivity of olefins of various
structure in hydrogenation over a silver catalyst. Ukr. khim.
zhur. 30 no.6:596-599 '64. (MIRA 18:5)

1. Institut fizicheskoy khimii imeni Pisarzhevskogo AN UkrSSR.

BELOUSOV, V.M.

Absorption spectra of silicate ceramics following irradiation. Izv.
vys. ucheb. zav.; fiz. 8 no.2:175-177 '65. (MIRA 18:7)

1. Tomskiy politekhnicheskiy institut imeni Kirova.

CHERTOV, V.M.; BELOUISOV, V.M.

Use of silica gel as a carrier for gas-liquid chromatography.
Ukr.khim.zhur. 31 no.2:171-174 '65.

(MIRA 18:4)

1. Institut fizicheskoy khimii im. L.V.Pisarzhevskogo AN UkrSSR.

BELOUSOV, V.M.; RUBANCHIK, M.Ya.; GERSHINGORINA, A.V.

Pulse method of studying the kinetics of reactions under conditions of the unsteady state of a catalyst. Ukr.khim. zhur. 31 no.5:444-449 '65. (MIRA 18:12)

1. Institut fizicheskoy khimii AN UkrSSR imeni Pisarzhevskogo. Submitted Febr. 10, 1964.

BELOUSOV, V.M.; GERSHINGORINA, A.V.

Method of preparing modified activated carbon for gas chromatography.
Ukr. khim. zhur. 31 no.6:633-635 '65. (MIRA 18:7)

1. Institut fizicheskoy khimii imeni Pisarzhevskogo AN UkrSSR.

ZHUCHKOV, P.A.; ABRAMOVICH, A.D.; BELOUSOV, V.N.

Effect of the hydraulic and thermal conditions of papermaking
machines on the rate of drying of paper sheets. Trudy LTITSB'
no.13:128-133 '64. (MIRA 18

BELOUSOV, V.N.; LIMANSKAYA, A.L.

For cleanliness of the air. Ogneupory 27 no.2:93 '62.
(MIRA 15:3)

1. Chasov-Yarskiy kombinat ogneupornykh izdeliy (for Belousov).
2. Konstantinovskiy ogneupornyy zavod "Krasnyy Oktyabr'" (for Limanskaya).

(Dust collectors)

131-1-3/14

AUTHOR: Belousov, V. N.

TITLE: Remodelling Two Blocks of Periodic Furnaces in the **Chasov-Yar Factory Imeni Ordzhonikidze** (Pereustroystvo dvukh blokov periodicheskikh pechey na Chasov-Yarskom zavode im. Ordzhonikidze)

PERIODICAL: Ogneupory, 1958, Nr 1, pp. 11 - 13 (USSR)

ABSTRACT: Each block of these furnaces consisted of 7 chambers, separated from each other by walls of 2 1/2 bricks. The usable area of one chamber was 15,2 m², the capacity - 47 m³. The furnaces were provided with semigas firings and step grates for the firing of Donets coal. The furnace pressure was done by low-pressure exhausters N 12 1/2 and 9 1/2, system "Sirokko". The remodelling of these furnaces into an annular gas furnace was performed in the following manner: The dividing walls of the chambers were taken apart and thus the 7 chambers of one block united to one furnace flue. 2 blocks in this manner gave an annular furnace as is to be seen in figure 1. The arrangement of the gas conduits is shown in figure 2. The annular gas furnace is fired with cold generator gas with a calorie content of 1250 - 1270 kcal/m³ and a gas pressure of 150 mm WS. The rise of temperature per hour is shown in a table and then the method of operation of the furnace is described. The re-

Card 1/2

131-1-3/14

Remodelling Two Blocks of Periodic Furnaces in the **Chesov-Yar Factory Ineni**
Ordzhonikidze

sult of the remodelling of the furnace: content of chamber enlarged to 51,5 m³ the output of furnace increased to 2800 t products per month with 1,5 % waste (before remodelling 2,100 t and 3,5 % waste). The specific fuel consumption was reduced from 290 to 142 kg. The working conditions for the laborers were improved. There are 2 figures, and 1 table.

AVAILABLE: Library of Congress
1. Furnaces-Remodelling

Card 2/2

BELOUSOV, V.N., arkhitektor

Principal construction problems of the city of New York. Gor.
khoz.Mosk. 36 no.1:46-48 Ja '62. (MIRA 16:1)
(New York (City)—Construction industry)

BELOUSOV, V.N., inzh.

~~The~~ The energy factor in increasing the vapor pressure for the
intensification of paper drying. Trudy LTITSBP no.8:146-152
'61. (MIRA 16:9)

(Paper--Drying)

ZHUCHKOV, P.A.; BELOUSOV, V.N.

Investigating the aerodynamic and psychometric conditions of the operation of the ventilation system of high-speed papermaking machines. Trudy LTITSBP no.13:134-139 '64.

(MIRA 18:2)

BELOUSOV, V.O. [Bielousov, V.O.], prof.

Serous toxicallergic meningitis in tuberculous children. Ped., akush.
i gin. 19 no.5:24-28 '57. (MIRA 13:1)

1. Kafedra pediatrii (zav. - prof. V.O. Belousov) Khar'kovskogo medi-
tsinskogo instituta (dir. - dots. I.P. Kononenko).
(MENINGES--TUBERCULOSIS)

VOLKOV, N.Z.; BELOUSOV, V.P.

Using a mining method in laying underground piping. Prom. stroi. 40
no.7:36-37 '62. (MIRA 15:7)

1. Trest Metallurgstroy, Tula.
(Tunneling) (Metallurgical plants--Equipment and supplies)

ALTSYBEREVA, A.I.; BELOUSOV, V.P.; OVTRAKH, N.V.; MORACHEVSKIY, A.G.

Phase equilibria and thermodynamic properties of the system
sec-butyl alcohol - water. Zhur. fiz. khim. 38 no.5, 1242-1247
My '64. (MIRA 18)12)

L. Leningradskiy gosudarstvennyy universitet imeni Zhdanova.
Submitted July 6, 1963.

BELOUSOV, V.P.; KOMAROV, Ye.V.; MORACHEVSKIY, A.G.

Thermodynamic properties of the system n-propyl alcohol - n-propyl acetate. Zhur.fiz.khim. 39 no.10:2499-2502 O '65.

(MIRA 18:12)

1. Leningradskiy gosudarstvennyy universitet imeni Zhdanova.
Submitted July 24, 1964.

BELOUSOV, V.P.; MORACHEVSKIY, A.G.

Thermodynamic properties of the system n-propyl acetate - water.
Zhur.fiz.khim. 39 no.11:2701-2703 N '65.

(MIRA 18:12)

L. Leningradskiy gosudarstvennyy universitet imeni A.A.Zhdanova.

BELOUSOV V.P.

AUTHORS: Morachevskiy, A. G., Belousov, V. P.

54-1-11/17

TITLE: The Study of Three-phase Equilibria in the Benzene-Ethyl Alcohol-Water System (Issledovaniye trekhfaznykh ravnovesiy v sisteme benzol-etilovyy spirt-voda)

PERIODICAL: Vestnik Leningradskogo Universiteta Seriya Fiziki i Khimii (Nr 1), 1958, Nr 4,

ABSTRACT: M. S., Vrevskiy (ref. 1) set up rules which determine the influence of the solution exercised by a change of temperature on the composition of the vapors and of the composition azeotropic mixtures. These problems were investigated by numerous authors and were extended to multicomponent systems. As yet, however, the rules governing the influence of temperature on the composition of the vapors of solutions able to dissociate and on the composition of heteroazeotropes remains unexplained (in the case of binary systems, if by composition of heteroazeotropes a composition of vapor is meant which is in equilibrium with two liquid layers, the two problems are identical. In ternary systems, however, these problems differ as to their respective significance).

Card 1/2

The Study of Three-phase Equilibria in
the Benzene-Ethyl Alcohol-Water System

54-1-11/17

The authors thank professor A.V. Storonkin for his kindness. The paper gives the result of the liquid-liquid vapor equilibrium in the system $C_6H_6-C_2H_5OH-H_2O$ at temperatures of 35, 45, 55 and 64°C. The composition of the ternary heteroazeotropes at different temperatures and pressures was determined. There are 6 figures, 6 tables, and 11 references, 6 of which are Slavic.

SUBMITTED: February 4, 1957

AVAILABLE: Library of Congress

1. Vapor compounds-Theoretical analysis

BELOUSOV, V.P.

AUTHORS:

Storonkin, A.V., Morachevskiy, A.G.,
Belousov, V.P.

54-10-2-9/16

TITLE:

The Effect of Temperature on the Composition of
Binary Heteroazeotropes (O vliyani temperature na sostav
binarnykh geteroazeotropov)

PERIODICAL:

Vestnik Leningradskogo Universiteta, Seriya fiziki i
khimii 1958, Vol. 10, Nr 2, pp. 94-100 (USSR)

ABSTRACT:

This paper deals with general equations which determine the changes in the composition of binary heteroazeotropes in the case of a change of temperature, and the correlation between the change of composition and the shape of the curve of the reciprocal solubility of the liquids is discussed. It is obvious that the question relating to the influence exercised by temperature upon the composition of the binary heteroazeotropes is identical with that of the influence of temperature upon the vapor composition of the binary stratifying solution. Therefore, this problem is best solved by means of equations which express the equilibrium of the three-phase systems (Ref 5). On the basis of the equation:

Card 1/4

The Effect of Temperature on the
Composition of Binary Heteroazeotropes

54-10-2-9/16

$$\frac{dx^{(3)}}{dT} = \frac{\tau_{31} - \tau_{32}}{\left(\frac{\partial \tau}{\partial x^2}\right)_{P,T} (x^{(2)} - x^{(1)})} = \frac{q_{31} - q_{32}}{T \left(\frac{\partial q}{\partial x^2}\right)_{P,T} (x^{(2)} - x^{(1)})}$$

the following conclusion may be drawn: If temperature rises, content of components in the binary heteroazeotrope increases which shows high concentration in that liquid layer during the condensation of which the greatest amount of heat is separated. This layer has a high differential molar evaporation temperature, and therefore the formulation given cannot be circumscribed accordingly. The problem of the influence exercised by temperature upon the change of composition of the binary heteroazeotrope can also be solved by another method, which also offers certain advantages. Let it be assumed that the partial molar temperature of the evaporation of one of the components is higher in all compositions of the solutions than that of others. A - component with higher evaporation temperature, L_A - its evaporation temperature, x_A - molar part. In view of such a condition it is possible to name 4 variants of the dependence of the reciprocal solubility of the

Card 2/4

The Effect of Temperature on the
Composition of Binary Heteroazeotropes

54-10-29 /16

components upon temperature:

- 1.) With rising temperature reciprocal solubility increases:
The A-component content increases in the layer with the
lowest A-content.
- 2.) With rising temperature reciprocal solubility diminishes;
the A-component content increases in the layer having the
highest A-content.
- 3.) With rising temperature the content of A in both layers in-
creases.
- 4.) With rising temperature the content of A in both layers
diminishes.

These variants of the solubility curves are shown (fig.2). It is possible that such systems exist in which the correlation of the partial molar temperature of the evaporation of components will be different in different layers; thus, for one layer it may be $L_A > L_B$, and for the other $L_B > L_A$. However, for such cases, which are apparently very rare, it is practically of no interest

Card 3/4

The Effect of Temperature on the
Composition of Binary Heteroazeotropes

54-10-2-9 /16

to connect the change of heteroazeotropic composition with the temperature values of the evaporation of components, though discussions, such as the one mentioned above, are, of course, possible also in this case. There are 2 figures, and 9 references, 6 of which are Soviet.

SUBMITTED: December 25, 1957
AVAILABLE: Library of Congress

1. Binary heteroazeotropes—Temperature factors—Theory
2. Binary heteroazeotropes—Solubility—Temperature factors

Card 4/4

Belousov, V.P.

MORACHEVSKIY, A.G.; BELOUSOV, V.P.

Study of three phase equilibria in the system benzene-ethyl alcohol-
water [with summary in English]. Vest. LGU 13, no. 4: 117-125 '58.
(Benzene) (Ethyl alcohol) (MIRA 11:4)
(Phase rule and equilibrium)

S/054/61/000/001/008/008
B117/B203

AUTHOR: Belousov, V. P.

TITLE: Heats of mixing of liquids. 1. Design of the calorimeter and determination of heats of mixing of n-propyl alcohol and water

PERIODICAL: Vestnik Leningradskogo universiteta. Seriya fiziki i khimii, no. 1, 1961, 144-148

TEXT: The author describes the design of a calorimeter for determining the mixture temperatures of liquids at temperatures up to the boiling point, and gives experimental data with respect to the mixture temperature in the system n-propyl alcohol/water. This paper is part of an extensive investigation of the effect of temperature on the thermodynamic properties of two- and polyphase systems. Fig. 1 schematically shows the calorimeter built by the author. It works on a similar principle as the calorimeter built by S. M. Skuratov (Ref. 1: S. M. Skuratov. Kolloidn. zhurn., 9, no. 2, 133, 1947), but has a different design. It consists of: 1 = stainless, thin-walled steel cylinder of 30 mm diameter and 50 mm height; ✓

Card 1/6

Heats of mixing of liquids...

S/054/61/000/001/008/008
B117/B203

2 = cover; 3 = opening in the cover; 4 = table; 5 = table leg; 6 = ampul; 7 = thin-walled bell with 8 = opening for filling the liquid; 9 = Manganin wire heater ($R = 36$ ohms); 10 = glass tube with heater wires; 11 = point for breaking the ampuls. The mixing vessel is set into a copper bucket (13); the latter is attached to three Textolite supports (14) in a thin-walled cylinder (15); the thermobattery (12) situated between bucket and cylinder consists of five copper-Constantan elements. The brass cylinder (15) is hermetically sealed on top with a cover (16) with polyethylene inset (17). The brass tube (25) guides the wires and fixes the cylinder to the cover of the adiabatic casing. Interposition of bearings (18) between tube and cylinder permits smooth rotation of the latter. The cylinder (15) is inserted in an adiabatic casing. The latter is a copper vessel (23) holding about 7 l of petroleum or, at temperatures above 50°C , of oil. On the bottom, there is a large-area nichrome heater (22). A 2000-rpm propeller mixer (19) is attached to the Textolite cover of the casing, as well as a second heater (20), and a cooler of four thin aluminum tubes (21). In the cover, there are openings for the thermometer and for heat conductors. The casing is lined outside with a 10 cm thick insulating foam layer (24). The temperature of the casing was recorded with an

Card 2/6

Heats of mixing of liquids...

S/054/61/000/001/008/008
B117/B203

MMT-1 (MMT-1) heat conductor which was inserted in a bridge circuit with an M 25/3 (M 25/3) galvanometer. The sensitiveness of the galvanometer was 10^{-9} A/mm. The accuracy of measurement was $\pm 0.0005^{\circ}\text{C}$. When preparing the experiments, the filling of the mixing vessel is of special importance. The total volume of the two liquids did not exceed 4-5 ml. The determination of the mixing effect is generally started after adjusting the effect of temperature to zero under maintenance of adiabatic conditions. After establishing the temperature balance of calorimeter and casing, the temperature was read off, and the change in temperature ΔT_1 of the calorimeter due to the mixing process was determined. The correction with respect to the imperfectness of adiabatic conditions was 0.1 - 0.2 %. The second part of the experiment was the determination of the thermal capacity of the system by the passage of current through the calorimeter heater. A compensation circuit with an P 375 (R375) potentiometer was used to measure the amount of current passing through. To check the dependability of the method, the thermal capacity of a filled calorimeter was measured at varying heating rate (1°C in 5, 10, and 15 min). The error of determination of the thermal capacity did not exceed 0.5 %. The

Card 3/6

Heats of mixing of liquids...

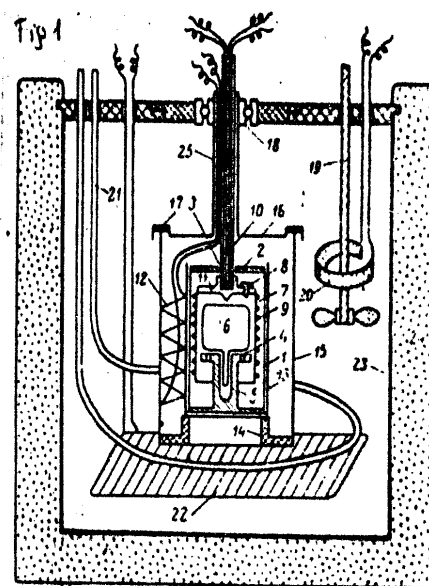
S/054/61/000/001/008/008
B117/B203 ✓

error of determination of ΔT_1 varied from 0.05 to 0.5 %, depending on T_1 (between 0.1 and 1°C). The error of prescription of the mixing proportion may attain 0.5 %. Deviations from adiabatic conditions during mixing due to local overheating may cause an error of 0.2 %. The maximum total relative error in determining the heat of mixing is ± 2 %. To check the method, heats of mixing were also measured in the system benzene/carbon tetrachloride. Results were in good agreement with dependable data of some other authors. Special tests showed that the adiabatic calorimeter described worked with unreduced accuracy even at elevated temperatures. The heats of mixing in the system n-propyl alcohol/water were measured at temperatures of 25°C, 50°C, and 75°C. Results are tabulated. A comparison of data with results found by E. Bose and A. Müller (Zs. phys. Chem., 58, 587, 1907) showed good agreement. The author thanks A. V. Storonkin, Professor, and A. G. Morachevskiy, Docent, for valuable hints. There are 4 figures, 1 table, and 10 references: 2 Soviet-bloc and 8 non-Soviet-bloc.

Card 4/6

Heats of mixing of liquids...

S/054/61/000/001/008/008
B117/B203



Card 5/6

Heats of mixing of liquids...

S/054/61/000/001/008/008
B117/B203

Legend to the table: x_{PrOH} denotes the molar alcohol fraction, ΔH the mixing enthalpy in calories per mole of the mixture.

25°C		50°C		75°C	
x_{PrOH}	$\Delta H, \text{ ккал/моль}$	x_{PrOH}	$\Delta H, \text{ ккал/моль}$	x_{PrOH}	$\Delta H, \text{ ккал/моль}$
0,027	-53,3	0,031	-18,2	0,008	-2,1
0,034	-69,3	0,043	-29,1	0,032	4,2
0,054	-78,7	0,082	-23,3	0,080	33,4
0,094	-91,8	0,099	-12,6	0,348	164
0,163	-65,7	0,206	30,0	0,506	186
0,262	-24,7	0,369	88,5	0,589	183
0,295	-19,4	0,466	104	0,665	177
0,349	-2,71	0,587	113	0,790	140
0,533	32,0	0,707	110	—	—
0,602	40,2	0,872	56,9	—	—
0,739	42,5	—	—	—	—

BELOUSOV, V.P.; ZHIGUNOV, I.S.; MORACHEVSKIY, A.G.

Heats of mixing of liquids. Part 2: Heats of mixing in binary systems n-propyl alcohol - n-propyl acetate, n-propyl acetate - water, and cyclohexane - methanol. Vest LGU 16 no.22:111-115 '61. (MIRA 14:11)

(Systems (Chemistry)) (Heat of mixing)

BELOUSOV, V.P.; MORACHEVSKIY, A.G.; STORONKIN, A.V.

Heat of mixing liquids. Part 3: Heat of mixing for ternary systems.
Experimental study of heats of mixing in the ternary system n-propyl
alcohol - n-propyl acetate - water. Vest. LGU 17 no.4:96-105 '62.
(MIRA 15:3)

(Propyl alcohol)(Acetic acid)(Heat of mixing)

BELOUSOV, V.P.; SABININ, V.Ye.; DMITRIYEV, I.V.

Calorimeter for determining the integral heats of vaporization of liquid mixtures. Izv.vys.ucheb.zav.; khim. i khim.tekh. 7 no.2:335-340 '64. (MIRA 18:4)

1. Leningradskiy gosudarstvennyy universitet im. A.A.Zhdanova, kafedra teorii rastvorov.

BELOUSOV, V.P.; MORACHEVSKIY, A.G.

Thermodynamic properties of the binary system n-propyl alcohol -
water. Zhur. fiz. khim. 38 no.1:184-189 Ja'64. (MIRA 17:2)

1. Leningradskiy gosudarstvennyy universitet imeni Zhdanova.

BAKALOV, S.A.; BELOUSOV, V.P.; HRATSEV, L.A.; VODOLAZKIN, V.M.;
YEROSHENKO, V.N.; ZHUKOV, V.F.; LUBAN, S.A.; MARKIZOV, L.P.;
NADEZHDIN, A.V.; NOVIKOV, F.Ya.; PONOMAREV, V.D.; POTRASHKOV,
G.D.; ROZHDESTVENSKIY, S.I.; TROFIMOV, S.V.; FEL'DMAN, I.R.;
FOYGEL', D.O.; KHRUSTALEV, L.N.; CHURUKSAYEV, I.I.;
KONDRAT'YEVA, V.I., red.

[Theory and practice in the study of frozen ground in construc-
tion] Teoriia i praktika merzlotovedeniia v stroitel'stve. Mo-
skva, Nauka, 1965. 187 p. (MIRA 18:4)

1. Moscow. Nauchno-issledovatel'skiy institut osnovaniy i pod-
zemnykh sooruzheniy. Severnoye otdeleniye.

STORONKIN, A.V.; BELOUSOV, V.P.

Limitations imposed by the stability conditions on the nature
of the concentration dependence of thermodynamic functions of
mixing. Zhur. fiz. khim. 39 no. 1:174-177 Ja '65
(MIRA 19:1)

1. Leningradskiy gosudarstvennyy universitet imeni A.A. Zhdanova.
Submitted December 12, 1963.

BELOUSOV, V.R.; LEBEDEV, L.N.; POLYAK, G.I.

Simulation of turbines and their speed regulators. Izv. NIIFT
no.5:273-284 '60. (MIRA 14:1)
(Turbines--Electromechanical analogies)

BELOUSOV, V.R.; SADOVSKIY, Yu.D.

Device for measuring e.m.f. phase shifts in electrical system
models. Izv. NIPT no.6:277-283 '60. (MIRA 14:7)
(Electric network analyzers)
(Electronic measurements)
(Electric machinery, Synchronous)

1. BELOUSOV, V. S., Eng.
2. USSR (600)
4. Electric Power Plants
7. Competition in the quality of equipment repair and in lowering its costs, Elek. sta., 23, No. 11, 1952.

9. Monthly List of Russian Accessions, Library of Congress, February 1953. Unclassified.

1. BELOUSOV, V.S.
2. USSR (600)
4. Electric Power Plants
7. Competition "For outstanding quality and lower cost of repair of every unit of equipment." Eng. V.S. Belousov, Rab.energ. 3 no. 3, 1953.

9. Monthly List of Russian Accessions, Library of Congress, APRIL 1953, Uncl.

BELOUSOV, V.S., obshchiy red.; KRYZHANOVSKIY, V.A., obshchiy red.;
SHUKHER, S.M., red.; LARIONOV, G.Ye., tekhn.red.

[Operation of the Cherepets State-owned Regional Electric
Power Plant] Opyt ekspluatatsii Cherepetskoi GRES. Moskva,
Gos.energ.isd-vo, 1959. 302 p. (MIRA 12:9)
(Cherepets--Electric power plants)

BELOUSOV, V.V., red.; BELYAYEVSKIY, N.A., red.; BOGDANOV, A.A.,
red.; GARETSKIY, R.G., red.; GUBIN, I.Ye., red.; K
KROPOTKIN, P.N., red.; LEYTES, A.M., red.; MAZAKOVICH,
O.A., red.; MURATOV, M.V., red.; NIKOLAYEV, N.I., red.;
PAVLOVSKIY, Ye.V., red.; PEYVE, A.V., red.; PETRUSHEVSKIY,
B.A., red.; PUSHCHAROVSKIY, Yu.M., red.; SHEYNMANN, Yu.M.,
red.; SHTREYS, N.A., red.; YANSHIN, A.L., red.

[Problems of the comparative tectonics of ancient platforms;
materials] Voprosy sravnitel'noi tektoniki drevnikh platrofm;
materialy. Moskva, Nauka, 1964. 152 p. (MIRA 17:8)

SHOLPO, Viktor Nikolayevich; BELOUSOV, V.V., otv. red.

[Type and formation of folds in the shale-bearing part of Daghestan] Tipy i usloviia formirovaniia skladchatosti slantseвого Dagestana. Moskva, Izd-vo "Nauka," 1964. 167 p. (MIRA 17:6)

1. Chlen-korrespondent AN SSSR (for Belousov).

PEYVE, A.V., otv. red.; BELCUSEV, V.V., red.; GARETSKIY, R.G.,
red.; LEYTES, A.M., red.; PAVLOVSKIY, Ye.V., red.;
YANSHIN, A.L., red.

[Deformation of rocks and tectonics] Deformatsiia porod i
tektonika. Moskva, Nauka, 1964. 274 p. (Doklady sovetskikh
geologov. Problema 4)
(MIRA 17:10)

1. Natsional'nyy komitet geologov Sovetskogo Soyuz.

MURATOV, M.V., *otv. red.*; PUSHCHAROVSKIY, Yu.M., *red.*; KHAIN, V.Ye., *red.*; MAZAROVICH, O.A., *red.*; BELOUSOV, V.V., *red.*; BELYAYEVSKIY, N.A., *red.*; BOGDANOV, A.A., *red.*; GARETSKIY, R.G., *red.*; GUBIN, I.Ye., *red.*; KROPOTKIN, P.N., *red.*; LEYTES, A.M., *red.*; NIKOLAYEV, N.I., *red.*; PAVLOVSKIY, Ye.V., *red.*; PEYVE, A.V., *red.*; PETRUSHEVSKIY, B.A., *red.*; SHEYNMANN, Yu.M., *red.*; SHTREYS, N.A., *red.*; YANSHIN, A.L., *red.*

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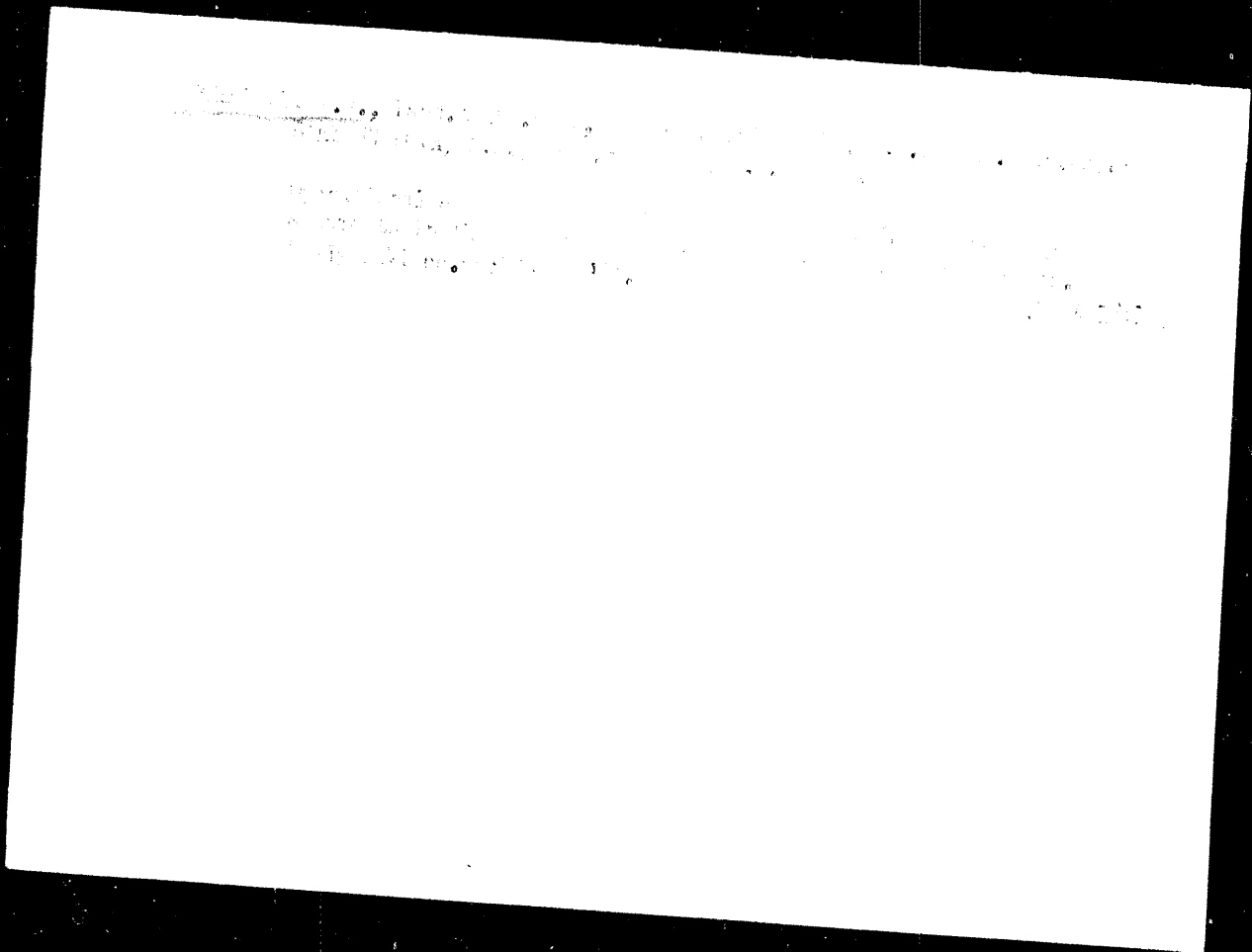
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GARETSKIY, R.G., otv. red.; YANSHIN, A.L. akademik, otv. red.;
BELOUSOV, V.V., red.; BELYAYEVSKIY, N.A., red.; BOGDANOV,
A.A., red.; GUBIN, I.Ye., red.; KROPOTKIN, P.N., red.;
LEYTES, A.M., red.; MAZAROVICH, O.A., red.; MURATOV, M.V.,
red.; NIKOLAYEV, N.I., red.; PAVLOVSKIY, Ye.V., red.; PEYVE,
A.V., red.; PETRUSHEVSKIY, B.A., red.; PUSHCHAROVSKIY, Yu.M.,
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1. Predsedatel' Mezhdunarodnogo geofizicheskogo Komiteta (MIRA 19:1)
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2. Vitse-prezident Spetsial'nogo (mezhdunarodnogo) kom. teta
Mezhdunarodnogo goda spokoynogo Solntsa i predsedatel' sovetsskoy
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(Heating)

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33 J1'55. (MLRA 8:12)

(Heating from central stations)

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gibberellins, and the gibberellins are converted to gibberellins.

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Vod.i san.tekh. no.9:32 D '55.

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(Heating pipes)

(MLRA 9:3)

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REPIN, Nikolay Nikolayevich, kand.tekhn.nauk; SVESHNIKOV, I.P., kand.tekhn.
nauk, retsenzent; ~~BELUSOV, V.V.~~, kand.tekhn.nauk, retsenzent;
KAZ'MIN-BALASHOV, A.I., inzh. nauchnyy red.; SMIRNOVA, A.P., red.
izd-va; TOKER, A.M., tekhn.red.

[Plumbing] Sanitarno-tekhnicheskie ustroistva zdaniy. Izd. 2-oe,
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BELOUSOV, V.V., inzh.red.; MUNTZ, A.F., red.; TOME, A.N., tekhn.red.

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for production and auxiliary buildings of industrial installations]
Ukazaniia po proektirovaniu otopeniia i ventilatsii proizvodstven-
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nyy red.; NINEMYAGI, D.K., red.isd-va; MEDVEDEV, L.Ya., tekhn.red.;
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(Heating)

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MIZHERICHER, A.M., inzh., red.; STRASHNYKH, V.P., red.;
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Classification

"Fundamental Problems of Folding Mechanism," V. V. Belousov (with participation of I. V. Kirillova, E. A. Rozanovoy, A. V. Goryacheva), 26 pp

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Plastic mechanism of folding depends more on the distribution of plastic rocks than on harder layers. Because the flow of different rock layers varies in intensity, the material presses out to form the crest and troughs of a fold. The greater the plasticity, the deeper the fold. Such deformation

produces cleavage of several types: main cleavage (parallel to axial surfaces of the fold); axial cleavage (converging along antiforms); the S-like cleavage (curving of main cleavage); dynamic (curving of rock layers); and transverse cleavage. The degree of the deformation plasticity and cleavage concentration results in the formation of peaks and depressions.

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180762

USSR/Geophysics - Tectonics

Mar/Apr 91

"Problem of the Earth's Structure and Its Evolution," V. V. Belousov, Geophys Inst, Acad Sci USSR

"Iz Ak Nauk, Ser Geog i Geofiz" No 2, pp 4-16

Proposes geotectonic data points to differentiation (stratification) of substances, as well as to main depth process that set the earth's development. Introduces "multilayer" differentiation and attempts to explain, with aid of this representation, basic peculiarities of the earth's tectonic history. Submitted by Acad O. Yu. Schmidt.

180762

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USSR/Geophysics - Terrestrial Crust

Sep 51

"Problems of Structure and the Development of the
Terrestrial Crust," V. V. Belousov

"Priroda" No 9, pp 21-32

Tectonic motion of terrestrial crust proceeds
according to definite laws and combinations. De-
scribes the history of formation of the Earth's
crust. Separation of continents and oceans seems to be
not directly connected to geosynclines and plat-
forms, although some yet undetected interactions
should exist.

211T58

BELOUSOV, V. V.

176T42

USSR, Geophysics - Earth's Structure

Jan/Feb 51

"Problems of the Earth's Internal Structure and Development," V. V. Belousov, Geophys Inst, Acad Sci USSR

"In Ak Nauk SSSR, Ser Geog i Geofiz" Vol XV, No 1, pp 3-19

Consists of 2 parts: First is devoted to fundamental laws governing development of structure of earth's crust as established by geotectonics. Second analyzes problems of structure of terrestrial depths and of processes occurring in earth's core.

176T42

U-1709, 27 Feb 1952

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Structure of the Eastern Alps in connection with general tectonic
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[Tectonic faults, their types, and origin] Tektonicheskie rasryvy, ikh tipy
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USSR/Geophysics - Seismology

Sep/Oct 52

"Brief Survey of the Seismicity of the Caucasus in Comparison With Its Tectonic Structure," V. V. Belousov, I. V. Kirilova, A. A. Sorokiy, Geophysics Inst, Acad Sci USSR

"Iz Ak Nauk SSSR, Ser Geofiz" No 5, pp 3-9

Authors conclude that epicenters of earthquakes are coordinated to zones sepg highlands and lowlands and also to transverse belts coinciding with the transverse anticlinal folds in the

226660

General structure of the Caucasus. Western Caucasus has less seismic activity than eastern Caucasus. Received 17 Jun 52.

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"Tectonic fractures, their types and mechanics of formation", Trudy Geofizichesk. in-ta
AN SSSR, No 17 (144), 1952.